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# 10



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GUIDE TO

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for example a school psychologist is nearly finished with an sld evaluation in spite of this she is obligated to visit the classroom during ongoing instruction and make note of the student s behavior see table 8 1 in this school district the idea observation requirement is delegated to the school psychologist

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web this experimental approach required the introduction of a new unit for amount of substances the mole which remains indispensable in modern chemical science the mole is an amount unit similar to familiar units like pair dozen gross etc it provides a specific measure of the number of atoms or molecules in a bulk sample of matter

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web the amount in moles of pb  $3.04 \times 10^3$  significant figures iii the amount in moles of lead produced iv mass of lead produced to 3 significant figures b 35.61 g of tin reacts with exactly 42.60 g of chlorine  $Cl_2$  to form 78.21 g of tin iv chloride  $SnCl_4$  i

calculate the number of moles of tin chlorine and tin chloride a r values

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