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web the first two equations have been balanced for you alpha particle α 2 4 beta particle β gamma ray γ proton p neutron n 0 positron β^{+} 0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation ${}_{137}^{55}\text{Cs} \rightarrow {}_{137}^{56}\text{Ba} + x$ because the mass number is the same on both sides of the

equation x must have a

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web identify the missing particle in the following nuclear reaction ${}^1_0\text{n} + {}^{235}_{92}\text{u} \rightarrow {}^2_{10}\text{n} + {}^{137}_{52}\text{te} + 4 {}^2_2\text{he} + 97 {}^1_0\text{n}$

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