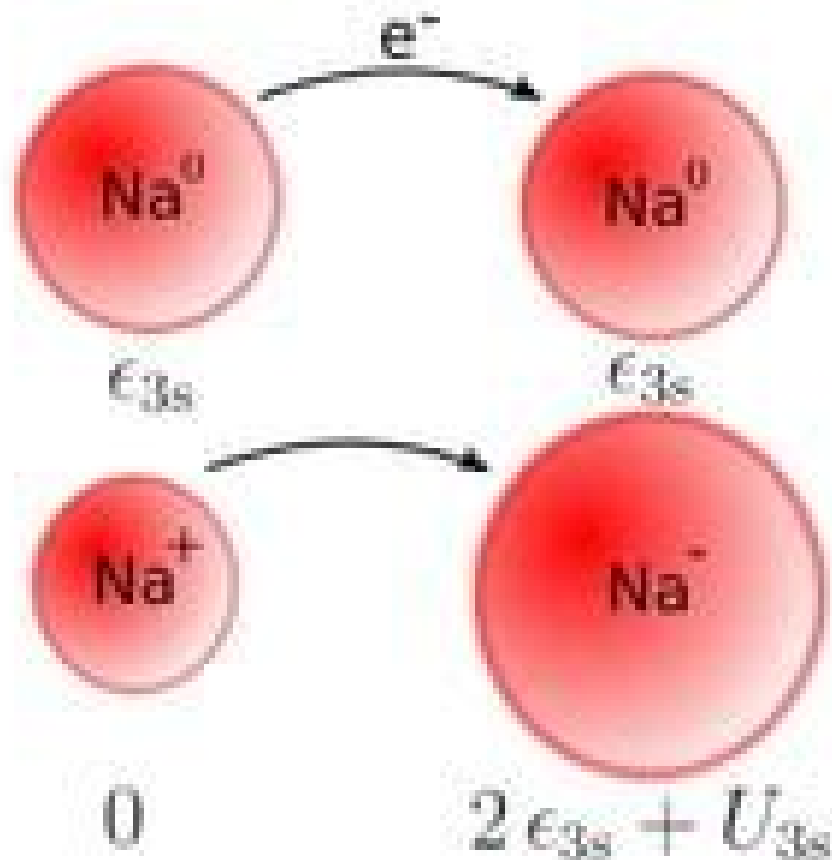


Importance of electronic correlations



- Hopping of electrons \rightarrow **kinetic energy gain.**
- Doubly occupied atomic site \rightarrow **Coulomb energy cost.**

- At small separations K.E. gain favors metallic behavior.
- At large separations, hopping of electrons are not energetically favorable.
- e-e interactions produce an insulating behavior.

$$U_{3s} = \int d^3r \int d^3r' |\phi_{3s}(\mathbf{r})|^2 \frac{e^2}{|\mathbf{r} - \mathbf{r}'|} |\phi_{3s}(\mathbf{r}')|^2$$

Correlations In Electronic And Atomic Fluids

JL Elias



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