

1. Draw Lewis structures, predict the molecular geometry, and give the bond angles for each of the following compounds:

a. Boron trifluoride,  $\text{BF}_3$

b. carbonate ion,  $\text{CO}_3^{2-}$

c. Sulfate ion,  $\text{SO}_4^{2-}$

d. Methane,  $\text{CH}_4$

e. Oxygen difluoride,  $\text{OF}_2$

f. Nitrogen trifluoride,  $\text{NF}_3$

g. Hydrogen cyanide,  $\text{HCN}$

2. For each of the Lewis structures you drew in question 1, indicate the hybridization of the central atom.

a. Boron trifluoride,  $\text{BF}_3$

b. carbonate ion,  $\text{CO}_3^{2-}$

c. Sulfate ion,  $\text{SO}_4^{2-}$

d. Methane,  $\text{CH}_4$

e. Oxygen difluoride,  $\text{OF}_2$

f. Nitrogen trifluoride,  $\text{NF}_3$

g. Hydrogen cyanide,  $\text{HCN}$

2.7

# Boronnitrogen Compounds

**Anton Meller**



## **Boron-nitrogen Compounds:**

Boron-Nitrogen Compounds Kurt Niedenzu, J. W. Dawson, 2012-12-06 Although the chemistry of boron is still relatively young it is developing at a pace where even specific areas of research are difficult to compile into a monograph Besides the boron hydrides boron nitrogen compounds are among the most fascinating derivatives of boron Nitrogen compounds exist in a wide variety of molecular structures and display many interesting properties The combination of nitrogen and boron however has some unusual features that are hard to match in any other combination of elements This situation was first recognized by ALFRED STOCK and it seems proper to pay tribute to his outstanding work in the area of boron chemistry One should realize that about forty years ago STOCK and his coworkers had to develop completely new experimental techniques and that no guidance for the interpretation of their rather unusual data had been advanced by theoretical chemists In this monograph an attempt has been made to explore the general characteristics of structure and the principles involved in the preparation and reactions of boron nitrogen compounds It was a somewhat difficult task to select that information which appears to be of the most interest to inorganic and general chemistry since the electronic relationship between a boron nitrogen and a carbon carbon grouping is reflected in the organic character of many of the reactions and compounds

**Boron-Nitrogen Compounds** Kurt Niedenzu, John William Dawson, 1965 An Investigation of Some Boron-nitrogen Compounds Robert Charles Osthoff, 1952 **B Boron Compounds** Anton Meller, 2013-03-09 The present issue Volume 3a of Boron Compounds 4th Supplement of the Gmelin Handbook presents the description of boron nitride and a part of other boron compounds containing nitrogen Volume 3b to be published Later will complete the presentation of the boron nitrogen compounds and will also cover boron compounds containing fluorine Due to technical circumstances Volume 1 systems with hydrogen and Volume 2 systems with oxygen of this particular supplement will be published subsequently whereas Volume 4 boron compounds containing Cl Br I S Se and Te as well as a section containing carboranes has already been published ALL volumes of the 4th Supplement will be augmented by a formula index The IUPAC nomenclature is generally adhered to occasional abbreviations for compounds are explained in the text A positive sign for the chemical shifts of the NMR signals indicates a 1-13 downfield shift from the references usually internal CH Si for 6 H and 6 C and external 3-4-11 C H h0 BF for 6 B others being specified 2-5-3 In contrast to more recent publications but remaining consistent with the previous volumes of the Boron Series the iminoborane structure is written as B-N and not as B=N In fact the real electronic structure is more realistically symbolized by B=N but it seems reasonable to avoid NR in accordance with the commonly accepted iminoborane nomenclature **Boron-nitrogen Compounds, by Kurt Niedenzu and John W. Dawson** Kurt Niedenzu, 1965 *Boron: Binary boron-nitrogen compounds, B-N-C-heterocycles. Polymeric boron-nitrogen compounds (new suppl. ser. v. 13)*, 1974

**A Chemical and Physical Study of Boron Nitrogen Compounds** T. J. Miller, 1980 **B Boron Compounds** Anton Meller, 1991-11-19 The present issue Volume 3a of Boron Compounds 4th Supplement of the Gmelin Handbook presents the

description of boron nitride and a part of other boron compounds containing nitrogen Volume 3 b to be published Later will complete the presentation of the boron nitrogen compounds and will also cover boron compounds containing fluorine Due to technical circumstances Volume 1 systems with hydrogen and Volume 2 systems with oxygen of this particular supplement will be published subsequently whereas Volume 4 boron compounds containing CL Br I S Se and Te as well as a section containing carboranes has already been published ALL volumes of the 4th Supplement will be augmented by a formula index The IUPAC nomenclature is generally adhered to occasional abbreviations for compounds are explained in the text A positive sign for the chemical shifts of the NMR signals indicates a 1 13 downfield shift from the references usually internal CH Si for 6 H and 6 C and external 3 4 11 C H h0 BF for 6 B others being specified 2 5 3 In contrast to more recent publications but remaining consistent with the previous volumes of the Boron Series the iminoborane structure is written as B N and not as B N In fact the real electronic structure is more realistically symbolized by B N but it seems reasonable to avoid NR in accordance with the commonly accepted iminoborane nomenclature

*B Boron Compounds* Anton Meller,1991-11-19 The present issue Volume 3a of Boron Compounds 4th Supplement of the Gmelin Hand book presents the description of boron nitride and a part of other boron compounds containing nitrogen Volume 3 b to be published Later will complete the presentation of the boron nitrogen compounds and will also cover boron compounds containing fluorine Due to technical circumstances Volume 1 systems with hydrogen and Volume 2 systems with oxygen of this particular supplement will be published subsequently whereas Volume 4 boron compounds containing CL Br I S Se and Te as well as a section containing carboranes has already been published ALL volumes of the 4th Supplement will be augmented by a formula index The IUPAC nomenclature is generally adhered to occasional abbreviations for compounds are explained in the text A positive sign for the chemical shifts of the NMR signals indicates a 1 13 downfield shift from the references usually internal CH Si for 6 H and 6 C and external 3 4 11 C H h0 BF for 6 B others being specified 2 5 3 In contrast to more recent publications but remaining consistent with the previous volumes of the Boron Series the iminoborane structure is written as B N and not as B N In fact the real electronic structure is more realistically symbolized by B N but it seems reasonable to avoid NR in accordance with the commonly accepted iminoborane nomenclature

**The Synthesis and Characterization of Several Boron-nitrogen Compounds** Brian Washburn,1973      **Boron-nitrogen Chemistry** Kurt Niedenzu,1964      **Mode of Formation and Reactions of Certain Organic Boron-nitrogen Compounds** Habibkhan Rahimkhan Pathan,Northern Polytechnic,1967

**Synthesis of New Boron-Nitrogen Compounds and Oligosilanes** Fabian Georg Müller,2019      **Binary Boron-Nitrogen Compounds, B-N-C-Heterocyclics. Polymeric Boron-Nitrogen Compounds** Leopold Gmelin,Richard Josef Meyer,Gmelin Institut für Anorganische Chemie und Grenzgebiete in der Max-Planck-Gesellschaft zur Förderung der Wissenschaften,Deutsche Chemische Gesellschaft,1974      **A Study of Organic Boro-nitrogen Compounds** Arden Richard Johnson,1911      **Synthesis and Spectroscopic Studies of Sigma-bonded Boron-nitrogen Compounds** Clifford Dale

Miller,1970 Boron and Nitrogen, Boron and Fluorine Anton Meller,1992-08-24 The volume presents compounds of boron with nitrogen and with fluorine and updates the preceding supplement series The majority of the boron nitrogen compounds includes amine boranes more than 250 compounds pyrazaboles and nitrogen substituted boron cations Emphasis is placed also on ammonium borates and on three and four coordinate boron nitrogen compounds containing B O bonds In the fluorine part special attention was given to BF<sub>3</sub> and to the tetrafluoroborates As many as 500 recent publications on applications of BF<sub>3</sub> and its etherate have been evaluated Ions radicals and neutral compounds containing only B F and partly also H or O are discussed Moreover linear and cyclic organylfluoroboranes and aminofluoroboranes and their adducts are treated Peroxy species are found among the OxBF<sub>y</sub> group compounds with four coordinate boron **Nuclear Magnetic Resonance**

**Studies of Boron-nitrogen Compounds** Akemi Saji,1961 **Some Reactions of Boron-nitrogen Compounds** Peter John BRATT,1972 **B Boron Compounds** Anton Meller,1991-11-19 The present issue Volume 3a of Boron Compounds 4th Supplement of the Gmelin Hand book presents the description of boron nitride and a part of other boron compounds containing nitrogen Volume 3 b to be published Later will complete the presentation of the boron nitrogen compounds and will also cover boron compounds containing fluorine Due to technical circumstances Volume 1 systems with hydrogen and Volume 2 systems with oxygen of this particular supplement will be published subsequently whereas Volume 4 boron compounds containing Cl Br I S Se and Te as well as a section containing carboranes has already been published ALL volumes of the 4th Supplement will be augmented by a formula index The IUPAC nomenclature is generally adhered to occasional abbreviations for compounds are explained in the text A positive sign for the chemical shifts of the NMR signals indicates a 1 13 downfield shift from the references usually internal CH Si for 6 H and 6 C and external 3 4 11 C H h<sub>0</sub> BF for 6 B others being specified 2 5 3 In contrast to more recent publications but remaining consistent with the previous volumes of the Boron Series the iminoborane structure is written as B N and not as B N In fact the real electronic structure is more realistically symbolized by B N but it seems reasonable to avoid NR in accordance with the commonly accepted iminoborane nomenclature

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