



## Introduction

Gasoline, perfume, alcohol, sugar, oil, and even all sorts of plastics; what something in common among these products? They are all chiefly made of organic compounds. There are many different types of organic compounds, but all have carbon as their principal constituent atom. These carbon atoms form a carbon skeleton or carbon backbone that has other bonded atoms such as H, N, O, S, and the halogens (F, Cl, Br, and I). And, this major type of substance is being studied in Organic chemistry. It describes the structures, properties, preparation, and reactions of a vast array of molecules that we call organic compounds.

Nowadays, people are getting health-conscious. Making them to resort on "organic products". But this frequently heard term which describes or refers to substances that are "natural" needs a little clarification. This is probably a result of the notion of early scientists that all organic compounds came from living systems and possessed a "vital force". However, chemists learned over 170 years ago that this is not the case. Organic compounds are major components of living systems, but chemists can make many of them in the laboratory from substances that have no direct connection with living systems.

Your journey through organic chemistry will be a challenging task for sure being a Science students. You will be studying about writing and drawing organic compounds, identifying functional groups, and a lot of challenging tasks to explore the world of organic compounds. Enjoy the journey ahead!

## Pre-Assessment

*Choose the letter of the correct answer.*

1. What are true about organic compounds?
  - I. Organic compounds contain calcium.
  - II. Organic compounds contain carbon.

# Advanced Module Q1

**Jian Lee**



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