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# Coal Desulfurization

**A Gutmann**



## Coal Desulfurization:

**Coal Desulfurization** S. K. Kawatra, Timothy C. Eisele, 2001 This book was written to provide a comprehensive survey of the current state of the art information in coal preparation with particular emphasis on coal desulfurization The primary audience for this book will be practising coal preparation engineers who need complete information about all of the coal preparation and desulfurization technologies that are available now or that may be available in the future It will also be valuable for coal researchers who need details and comparative data for cutting edge technologies that are still under development The main emphasis is on physical coal preparation but chapters also include chemical and biological technologies that are under development but not yet used in industrial practice Along with the successful technologies also included are details of processes and techniques that were attempted but were subsequently abandoned along with discussions of the reasons they were abandoned Coal Desulfurization Robert Allen Meyers, 1977 Coal Desulfurization Robert A. Meyers, Design Principles for a Coal Desulfurization Process with Iron Sulfides as in Situ Catalysts, 1986

Low Temperature Processes for Coal Desulfurization Eduardo Mendizabal, 1976 **Coal Desulfurization Prior to Combustion** Robert C. Eliot, 1978 **Desulfurization of Hot Coal Gas** Aysel T. Atimtay, Douglas P. Harrison, 1998-08-24 Proceedings of the NATO Advanced Study Institute on Desulfurization of Hot Coal with Regenerable Metal Oxide Sorbents New Developments held in Kusadasi Turkey July 1996 *A NOVEL APPROACH TO CATALYTIC DESULFURIZATION OF COAL.*, 1997 Remarkably mild conditions have been discovered for quantitative sulfur removal from dibenzothiophene and other organosulfur systems using relatively cheap elemental sodium The project objectives are 1 Optimize the coal desulfurization reaction with respect to time temperature coal type and the R groups including R H and also on extraction impregnation and sonication conditions 2 Optimize the conditions for the HDS reaction which allows the PR3 to function as an HDS catalyst for coal with respect to R group temperature pressure H2 gas flow rate and inert solvent presence 3 Determine the products and the pathway of the novel redox reaction that appears to quantitatively remove sulfur from dibenzothiophene DBT when R Bu when FeCl3 is used as a catalyst 4 Impregnate sulfur laden coals with Fe sup 3 to ascertain if the PR3 desulfurization rate increases 5 Determine the nature of the presently unextractable phosphorus compounds formed in solid coals by PR3 6 Explore the efficacy of PR3 Fe sup 3 in removing sulfur from petroleum feedstocks heavy ends whether solid or liquid coal tar and discarded tire rubber 7 Explore the possibility of using water soluble PR3 compounds and Fe sup 3 to remove sulfur from petroleum feedstocks and heavy ends in order to remove the SPR3 and Fe sup 3 catalyst by water extraction for subsequent HDS of the SPR3 and 8 Explore the possibility of using solid supported PR3 compounds plus Fe sup 3 catalyst to remove sulfur from petroleum feedstocks and heavy ends in order to keep the oil and the SPR3 formed in the reaction in easily separable phases Coal Desulfurization Thomas D. Wheelock, American Chemical Society. Division of Fuel Chemistry, 1977 Based on a symposium sponsored by the Division of Fuel Chemistry at the 173rd

meeting of the American Chemical Society *Removing Sulfur from Coal*, 1990 *Advanced Development of Fine Coal Desulfurization and Recovery Technology. Annual Technical Progress Report, October 1 1976--September 1977. [54 References]*, 1978 Work has been carried out to develop and improve several promising methods for desulfurizing and recovering fine size coal These methods included froth flotation selective oil agglomeration pelletization and a chemical desulfurization process which involves leaching fine coal with a hot dilute solution of sodium carbonate containing dissolved oxygen or air under pressure This process is an oxydesulfurization process which utilizes an alkaline solution instead of an acidic solution for leaching coal The results of experiments with a small autoclave showed that more sulfur both organic and inorganic was removed from several high sulfur bituminous coals when a dilute solution of alkali was used than when a dilute acid solution was used for leaching under the same conditions of time temperature and oxygen partial pressure On the other hand the heating value recovery was somewhat higher when acidic solutions were used Also it was shown that pure oxygen is more effective than air even though the oxygen partial pressure was kept the same for oxydesulfurization of coal in either acidic or alkaline solutions In addition it was found that the removal of organic sulfur by alkaline solutions increased as the oxygen partial pressure was raised Although small concentrations of alkali were beneficial large concentrations were detrimental In order to improve the physical separation of coal and pyrite particles by froth flotation or oil agglomeration various surface chemical treatments were tested The most successful treatment involved treating the coal or pyrite particles with a warm dilute alkaline solution which was aerated to provide mildly oxidizing conditions *Biological Coal Desulfurization* Ming Le, 1991 Sulfur removal from coal before combustion is of considerable interest to avoid emission of oxides of sulfur SO<sub>x</sub> Chemical and physical methods have been developed for removing the inorganic sulfur component in coal This component is generally associated with pyrites FeS<sub>2</sub> However no commercial methods have been developed for the removal of the organic sulfur component This study describes microbial methods for removing organically bound sulfur from coal The specific goals of this study were to ascertain whether a robust biologically active population such as activated sludge biomass can remove organic sulfur from coal and to devise sulfur analytical methods that are less cumbersome than the accepted standards for screening research results Dibenzothiophene is used in isolating microorganisms that use sulfur as the sole source for growth It was found that about 55% sulfur can be removed by activated sludge in shaker flasks from a coal which was previously treated to remove inorganic sulfur IBC 108 A no cycling leaching reactor with activated sludge removed 50% sulfur from the same type of coal Almost 72% sulfur is removed by A i S D1 T3 2 and Ar i mixed cultures It was also found that about 25 to 29% sulfur can be removed by different cultures from a commercially prepared washed coal in which the pyritic and organic sulfur have not been altered The application of ion chromatography and atomic spectrometry in analyzing sulfate pyritic sulfur organic sulfur and total sulfur following ASTM extraction can achieve easier and quicker sulfur determinations without sacrificing accuracy and reproducibility **Coal Desulfurization by a Microwave Process**

,1981 A laboratory size flow reactor was developed and tested on coal samples up to 100 g batch A modified version of the applicator will increase sample size to 1000 g when run continuously it can handle 500 g to 1000 g minute Recent measurements on typical 1 4 in size particle size coal has validated previous results which show up to 50% sulfur removal mainly by converting pyrite to FeS which is shown to have strong magnetic properties and the spectral characteristics of pyrrhotite Subsequent crushing and physical separation steps such as float sink or magnetic separation further reduces the sulfur level and more effectively than conventional separation Crushing of coal to 30 to 100 mesh and further exposure to the microwave field can further reduce the total sulfur level An 80% to 90% removal is possible with the proper concentration of NaOH and or multiple exposures Results are presented for several important Pennsylvania and Kentucky coals The results evaluated in terms of pounds of sulfur per 106 Btu suggest that this method had the potential of meeting environmental requirements *Coal Desulfurization Using Microwave Energy* Industrial Environmental Research Laboratory (Research Triangle Park, N.C.),1978 *Coal Desulfurization* Thomas D. Wheelock,1977-01-01 *Coal Desulfurization During the Combustion of Coal/oil/water Emulsions : an Economic Alternative Clean Liquid Fuel : Final Report April 1983* United States. Department of Energy,Adelphi Research Center,1983 **Fossil Energy Update** ,1980 **COAL DESULFURIZATION: CHEMICAL AND PHYSICAL METHODS- BASED ON A SYMPOSIUM HELD AT THE 173RD MEETING OF THE AMERICAN CHEMICAL SOCIETY- ACS, DIVISION OF FUEL CHEMISTRY.** , *Energy Research Abstracts* ,1979 *COAL DESULFURIZATION. A BIBLIOGRAPHY.* ,

## Reviewing **Coal Desulfurization**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Coal Desulfurization**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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