

You must remember your formulas!!!

$D = M/V$,

$$M = D \times V, \quad V = M/D$$

1. Find the density of a substance whose mass is 1.8×10^7 mg, and it has a volume of 3875 cm^3 .

 2. Find the volume of 10 grams of mercury which has a density of 13.5 g/cm^3 . Show the formula you use.

 3. What mass of silver can be held in a container with a volume of 150 cm^3 if silver has a density of 10.5 g/cm^3 ?

 4. A certain block of metal has dimensions which are 2.0 cm wide, 2.0 cm deep, and 10.0 cm high. The balance shows it to have a mass of 320 grams. What is its density?

 5. Refer to number 4 above. What would be the density of the block of metal if it was cut in half? Explain your answer.
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D V

Josiah Willard Gibbs



D V:

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