

Chemistry Reference Sheet

Cations

Cation name	Formula
Sodium	Na ⁺
Potassium	K ⁺
Magnesium	Mg ²⁺
Calcium	Ca ²⁺
Iron (II)	Fe ²⁺
Iron (III)	Fe ³⁺

Cation name	Formula
Hydrogen	H ⁺
Copper (I)	Cu ⁺
Copper (II)	Cu ²⁺
Zinc (II)	Zn ²⁺
Aluminium	Al ³⁺
Ammonium	NH ₄ ⁺

Anions

Anion name	Formula
Fluoride	F ⁻
Chloride	Cl ⁻
Bromide	Br ⁻
Oxide	O ²⁻
Sulfide	S ²⁻
Nitride	N ³⁻
Phosphate	P ₄ ³⁻
Carbonate	C ₃ ²⁻

Anion name	Formula
Hydroxide	OH ⁻
Cyanide	CN ⁻
Acetate	CH ₃ COO ⁻
Bromate	BrO ₃ ⁻
Nitrate	NO ₃ ⁻
Sulfate	SO ₄ ²⁻
Carbonate	CO ₃ ²⁻
phosphate	PO ₄ ³⁻

Covalent prefixes

Number of atoms of an element	Prefix
1	mono-
2	di-
3	tri-
4	tetra-
5	penta-
6	hexa-
7	hepta-
8	octa-

*the prefix for one atom is only used for the second element

Common Compounds

Anion name	Formula
Ammonia	NH ₃
Carbon dioxide	CO ₂
Ethanol	C ₂ H ₅ OH
Hydrochloric acid	HCl
Hydrogen peroxide	H ₂ O ₂
Nitric acid	HNO ₃
Octane	C ₈ H ₁₈
Sodium hydroxide	NaOH
Sulfuric acid	H ₂ SO ₄
Water	H ₂ O

General Equations for Calculations

Density $d = \frac{m}{V}$ Heat energy $Q = mc\Delta T$

Error $\text{error} = \text{measured} - \text{theoretical}$

%Error $\text{Error} = \frac{\text{measured} - \text{theoretical}}{\text{theoretical}} \times 100$

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Yan Bai



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