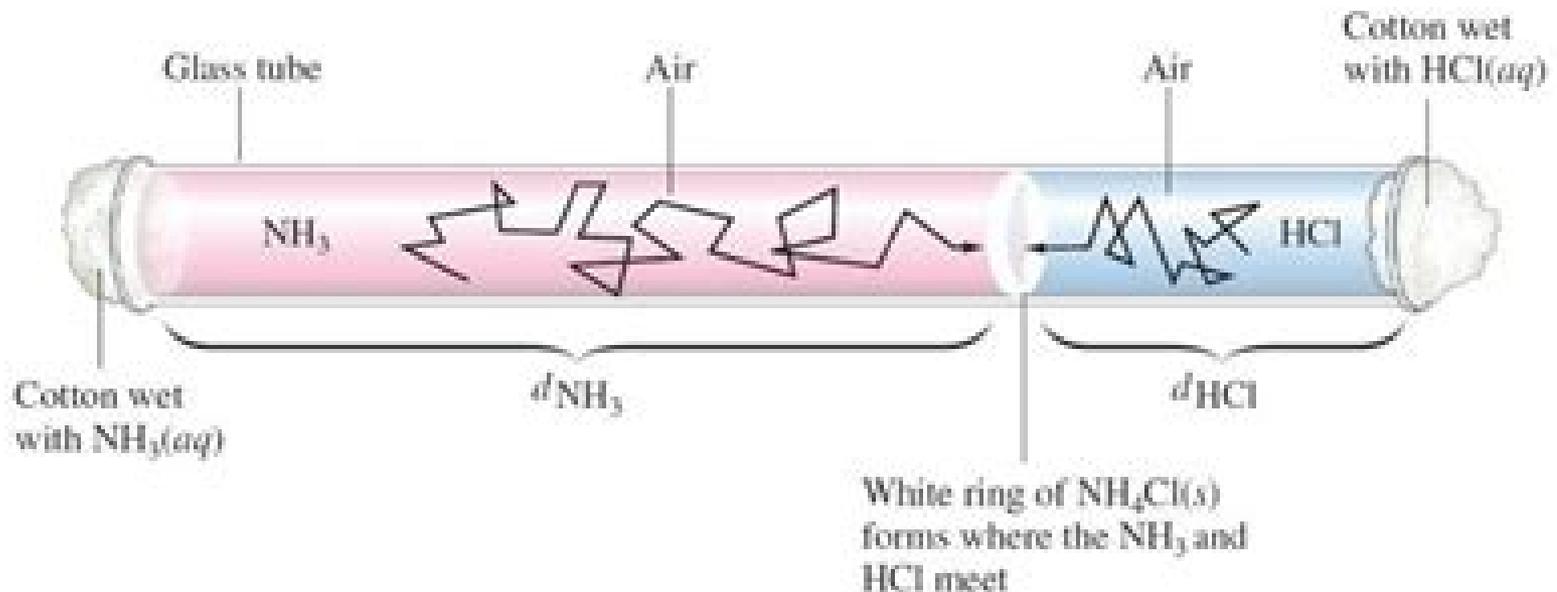


# Diffusion Reactions

- By similar arguments, the distance a molecule travels is inversely proportional to mass:

$$\frac{(\text{distance})_1}{(\text{distance})_2} = \frac{\sqrt{M_2}}{\sqrt{M_1}} \quad \longrightarrow \quad \frac{(\text{distance})_{\text{NH}_3}}{(\text{distance})_{\text{HCl}}} = \frac{\sqrt{M_{\text{HCl}}}}{\sqrt{M_{\text{NH}_3}}} = \sqrt{\frac{36.5}{17}} = 1.5 \quad \text{Exp. 1.3}$$

- Example: reaction of  $\text{NH}_3$  with  $\text{HCl}$



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