

11. Write a brief description or definition of each of the following:

- (a) nucleon
- (b) α particle
- (c) β particle
- (d) positron
- (e) γ ray
- (f) nuclide
- (g) mass number
- (h) atomic number

Solution

(a) A nucleon is any particle contained in the nucleus of the atom, so it can refer to protons and neutrons. (b) An α particle is one product of natural radioactivity and is the nucleus of a helium atom. (c) A β particle is a product of natural radioactivity and is a high-speed electron. (d) A positron is a particle with the same mass as an electron but with a positive charge. (e) Gamma rays compose electromagnetic radiation of high energy and short wavelength. (f) Nuclide is a term used when referring to a single type of nucleus. (g) The mass number is the sum of the number of protons and the number of neutrons in an element. (h) The atomic number is the number of protons in the nucleus of an element.

13. Complete each of the following equations by adding the missing species:

- (a) ${}_{13}^{27}\text{Al} + {}_2^4\text{He} \longrightarrow ? + {}_0^1\text{n}$
- (b) ${}_{94}^{240}\text{Pu} + ? \longrightarrow {}_{96}^{242}\text{Cm} + {}_0^1\text{n}$
- (c) ${}_{92}^{238}\text{U} \longrightarrow ? + {}_{55}^{138}\text{Cs} + 4{}_0^1\text{n}$
- (d) ${}_{92}^{238}\text{U} \longrightarrow ? + {}_{55}^{135}\text{Cs} + 4{}_0^1\text{n}$

Solution

- (a) ${}_{13}^{27}\text{Al} + {}_2^4\text{He} \longrightarrow {}_{15}^{31}\text{P} + {}_0^1\text{n}$, (b) ${}_{94}^{240}\text{Pu} + {}_2^4\text{He} \longrightarrow {}_{96}^{242}\text{Cm} + {}_0^1\text{n}$, (c) ${}_{92}^{238}\text{U} + {}_2^4\text{He} \longrightarrow {}_{94}^{242}\text{Pu} + {}_0^1\text{n}$, (d) ${}_{92}^{238}\text{U} \longrightarrow {}_{55}^{133}\text{Rb} + {}_{37}^{105}\text{Sr} + 4{}_0^1\text{n}$

15. Write a balanced equation for each of the following nuclear reactions:

- (a) the production of ${}^{17}\text{O}$ from ${}^{14}\text{N}$ by α particle bombardment
- (b) the production of ${}^{14}\text{C}$ from ${}^{14}\text{N}$ by neutron bombardment
- (c) the production of ${}^{230}\text{Th}$ from ${}^{232}\text{Th}$ by neutron bombardment
- (d) the production of ${}^{239}\text{U}$ from ${}^{238}\text{U}$ by ${}^3_1\text{H}$ bombardment

Solution

- (a) ${}^{14}_7\text{N} + {}^4_2\text{He} \longrightarrow {}^{17}_8\text{O} + {}^1_1\text{H}$, (b) ${}^{14}_7\text{N} + {}^1_0\text{n} \longrightarrow {}^{14}_6\text{C} + {}^1_1\text{H}$, (c) ${}^{232}_{90}\text{Th} + {}^1_0\text{n} \longrightarrow {}^{230}_{90}\text{Th} + {}^2_0\text{n}$, (d) ${}^{238}_{92}\text{U} + {}^3_1\text{H} \longrightarrow {}^{239}_{93}\text{Np} + {}^0_{-1}\text{e}$

17. The mass of the atom ${}^{18}\text{F}$ is 18.99840 amu.

- (a) Calculate its binding energy per atom in millions of electron volts.
- (b) Calculate its binding energy per nucleon.

Solution

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Ying Liu



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1. Understanding the eBook Ctb Organic Chem 2e Lecture Succe
 - The Rise of Digital Reading Ctb Organic Chem 2e Lecture Succe
 - Advantages of eBooks Over Traditional Books
2. Identifying Ctb Organic Chem 2e Lecture Succe
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ctb Organic Chem 2e Lecture Succe
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ctb Organic Chem 2e Lecture Succe
 - Personalized Recommendations
 - Ctb Organic Chem 2e Lecture Succe User Reviews and Ratings
 - Ctb Organic Chem 2e Lecture Succe and Bestseller Lists
5. Accessing Ctb Organic Chem 2e Lecture Succe Free and Paid eBooks
 - Ctb Organic Chem 2e Lecture Succe Public Domain eBooks
 - Ctb Organic Chem 2e Lecture Succe eBook Subscription Services
 - Ctb Organic Chem 2e Lecture Succe Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Ctb Organic Chem 2e Lecture Succe Compatibility with Devices
 - Ctb Organic Chem 2e Lecture Succe Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ctb Organic Chem 2e Lecture Succe
 - Highlighting and Note-Taking Ctb Organic Chem 2e Lecture Succe
 - Interactive Elements Ctb Organic Chem 2e Lecture Succe
8. Staying Engaged with Ctb Organic Chem 2e Lecture Succe

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ctb Organic Chem 2e Lecture Succe
9. Balancing eBooks and Physical Books Ctb Organic Chem 2e Lecture Succe
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ctb Organic Chem 2e Lecture Succe
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ctb Organic Chem 2e Lecture Succe
- Setting Reading Goals Ctb Organic Chem 2e Lecture Succe
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ctb Organic Chem 2e Lecture Succe
- Fact-Checking eBook Content of Ctb Organic Chem 2e Lecture Succe
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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