

Chem reac.

Alcohol - R-CH₂OH - -nol

กรดอินทรีย์ - R-COOH - -noic acid

Ester - R-COO-R' - -yl -noate

Phenol - Benzene-OH -

Ether - R-O-R' - ด้านสั้นoxyด้านยาวane
และตำแหน่งที่ O สอดกับด้านยาว

e.g. 2-methoxypropane



Aldehyde - R-CHO - anal

Ketone - R-CO-R' - -an-x-one

x = ตำแหน่งที่ o สอดกับ c

Amine - R-NH₂ (primary), R-NH-R' (secondary), R-NR'-R'' (tertiary) -

-an-x-amine x = ตำแหน่งที่ NH₂ สอดกับ C

Amide - R-CONH₂ (primary), R-CONH-R' (N-substituted),

R-CONR'-R'' (N,N-substituted) -anamide

Structure	Common name	IUPAC name
	Phenol	Phenol (Benzenol)
	m-cresol	3-methylphenol
	Catechol	2-hydroxyphenol
	Resorcinol	3-hydroxyphenol
	Quinol	4-hydroxyphenol
	Picric acid	2,4,6-trinitrophenol

Alcohol

1. กรดอินทรีย์ + แอลกอฮอล์ = เอสเทอร์ + น้ำ -noic acid + -nol = -yl -noate + H₂O

2. โลหะ + แอลกอฮอล์ = เกลือแอลกอฮอล์ + H₂ 2Metal + 2-nol = 2Metal -oxide + H₂O

3. ออกซิเดชัน แอลกอฮอล์ -> แอลดีไฮด์ -> กรดอินทรีย์ R-CH₂OH -> RCHO -> RCOOH

กรดอินทรีย์

1. กรดอินทรีย์ละลายน้ำได้ไอออน RCOOH + H₂O = RCOO⁻ + H₃O⁺

2. โลหะ + กรดอินทรีย์ = เกลือกรดอินทรีย์ + H₂ 2Metal + 2-noic acid = 2Metal -noate + H₂

3. กรดอินทรีย์ + คาร์บอเนต = เกลือกรดอินทรีย์ + H₂O + CO₂ RCOOH + XCO₃/XHCO₃ = RCOOX + H₂O + CO₂

4. การสะเทิน กรดอินทรีย์ + เบส = เกลือกรดอินทรีย์ + H₂O -noic acid + XOH = metal -noate + H₂O

5. รีดิวซ์ กรดอินทรีย์ -> แอลกอฮอล์ RCOOH -> RCH₂OH ผ่านตัวเร่ง

6. แอลฟาไฮดรอกซิล OH สอดกรดอินทรีย์ตำแหน่งแอลฟา H₃C(Gamma)-CH₂(Beta)-CH₂(Alpha)-COOH

Ester

1. ไฮโดรไลซิส R-COO-R' + H₂O = R-COOH + R'-OH ตรงข้าม esterification xtyl ztanoate -> znoic acid + xanol

2. R-COO-R' + เบสแก่(XOH) = R-COOX + R'-OH, 2R-COOX + กรดกำมะถัน(H₂SO₄) = กรดอินทรีย์ + XSO₄

Phenol

1. 2phenol + Metal = 2phenol but replace H from OH with metal (metal Xoxide) + H₂

2. phenol + เบส(NaOH) = phenol but replace H from OH with metal + H₂O

3. Halogenation replace H in benzene with หมู่ 7 phenol + 3Br₂ = phenol with 3Br + 3HBr

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SJ Ball



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