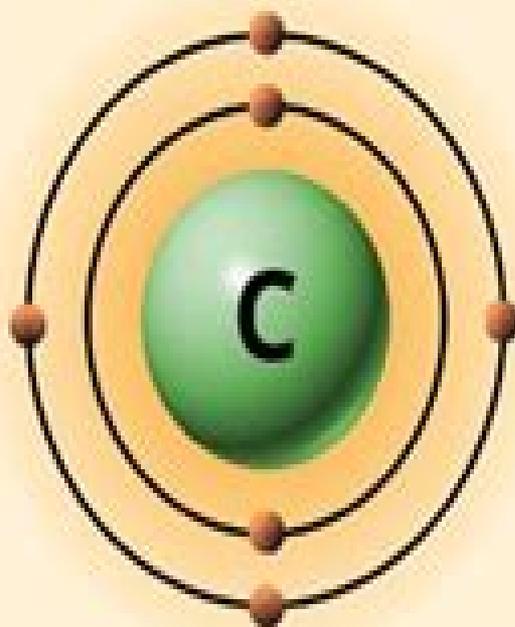
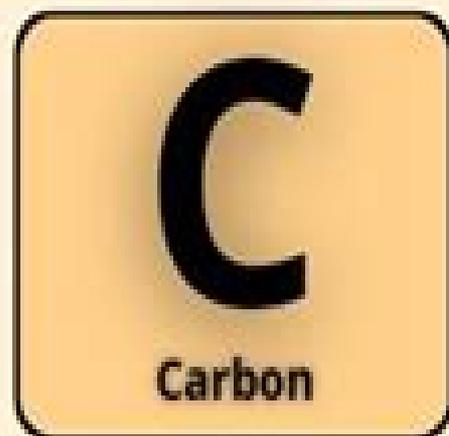


# Carbon element



Atomic number: **6**

Standard state: **Solid**

Group: **14**

Period: **2**

Atomic mass (u): **12.011**

Atomic radius (van der Waals): **170 pm**

Electron configuration: **[He] 2s<sup>2</sup> 2p<sup>2</sup>**

Electrons per shell: **2, 4**

1<sup>st</sup> Ionization energy: **17.423 eV**

Electronegativity (Pauling): **2.55**

Valency: **4**

Crystal structure: **Hexagonal**



# Carbon The Elements Group 1

**Leverett Mears**



## Carbon The Elements Group 1:

*Handbook of the Chemical Elements* Hermann Sicius, 2024-10-28 This reference work describes comprehensively compactly and precisely the history properties production and application of all elements of the periodic table Particular attention is paid to the chemical compounds of the elements which are also presented extensively This book contains 23 chapters each of which includes the elements in the form of subchapters of the eight main groups the first and second as well as the fourth to tenth subgroups the rare earth metals and the third subgroup as well as the actinides Finally there is an outlook on the as yet undiscovered elements of the eighth and ninth periods on alternative more environmentally friendly drives for motor vehicles such as batteries and fuel cells as well as on semiconductor technology i e areas that will continue to see increasing research activity in the future Whenever possible the author has always maintained the order of chalcogenides halides pnictogenides and other compounds when presenting the chemical compounds of the elements The introductory part which illuminates the history of the respective element often contains biographies of well known researchers whose creative periods range from the near past to the Middle Ages You will not only find portraits of chemists but also of nuclear physicists astronomers and medical doctors

**Product Engineering** James Wei, 2007-01-04 The current chemical engineering curriculum concentrates on process the efficient manufacturing in quantity of traditional chemical products such as ammonia and benzene However many chemical companies now invent and manufacture specialty products with particular properties such as pharmaceuticals cosmetics and electronic coatings and their employees need to know how to design the products as well as manufacture them James Wei a famous chemical engineer is writing this book to provide theories and case studies in product engineering the design of new useful products with desired properties The first section relates historical case studies of successful product invention and development by individuals and companies The second part of the book describes the toolbox of molecular structure property relations A desired product needs to have certain properties for example phase transition or thermal properties and the chemist must find or design a molecular structure with the required properties This section will instruct chemists in the analysis of structure and property information The third section is concerned with the next stage product research and design It will discuss improving the desired product by additives and blending among other strategies It will also cover future challenges in product engineering

**Outline of Lectures on the Non-metallic Elements and Their Compounds** Leverett Mears, 1896 **Proceedings of the Section of Sciences** Koninklijke Nederlandse Akademie van Wetenschappen. Afdeling Natuurkunde, 1918 [The Inorganic Compounds \(14,000\)](#) Adolf Law Voge, 1911 *A System of Mineralogy* James Dwight Dana, 1890 *Chemical Principles* Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses *Chemical Principles* helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually

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Inorganic Chemical Synonyms and Other Useful Chemical Data Elton Richmond Darling,1919

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