

# Acid Rain Information



Causes, Effects and Solutions to Acid Rain

## What is Acid Rain?

Acid rain is a result of air pollution. When any type of fuel is burnt, lots of different chemicals are produced. The smoke that comes from a fire or the fumes that come out of a car exhaust contains lots of invisible gases that can be even more harmful to our environment.

Power stations, factories and cars all burn fuels and therefore they all produce polluting gases. Some of these gases (especially nitrogen oxides and sulphur dioxide) react with the tiny droplets of water in clouds to form sulphuric and nitric acids. The rain from these clouds then falls as very weak acid - which is why it is known as "acid rain". The release of sulphur dioxide can also occur naturally when a volcano erupts.

- Sulphur dioxide and water form sulphurous acid ( $\text{H}_2\text{SO}_3$ )  
 $\text{SO}_2(\text{g}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_2\text{SO}_3(\text{aq})$
- Some oxides of nitrogen like nitrogen dioxide ( $\text{NO}_2$ ) react with water to form nitrous acid ( $\text{HNO}_2$ ) and nitric acid ( $\text{HNO}_3$ )  
 $2\text{NO}_2(\text{g}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{HNO}_2(\text{aq}) + \text{HNO}_3(\text{aq})$
- Carbon dioxide in the air can dissolve in rain water to form carbonic acid, ( $\text{H}_2\text{CO}_3$ ).
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$

## How acidic is Acid Rain?

Unpolluted rain would have a pH value of between 5 and 6. When the air becomes more polluted with nitrogen oxides and sulphur dioxide the acidity can increase to a pH value of 4.

# Acid Rain Information

**Ellis Mount**



## **Acid Rain Information:**

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