

doing sth, to make them pay for a mistake, etc: bring a libel - against sb
4 [u] fighting in war; his won killed in -
5 [u] events in a story, play, etc
6 [u] exciting events **(BGM) actions**
 speak louder than words what a person actually does means more than what they say they will do into action into operation out of action no longer working

activate /'æktɪveɪt/ *v* [T] start sth working

be active /'æktɪv/ *adj* **1** doing things; busy or energetic **2** (gram) of the verb form used when the subject of a sentence does the action, as in 'She drove the car.' **▶** the active (also active voice) *n* [sing] active **(2)** form of a verb **▶** actively *adv*

activist /'æktɪvɪst/ *n* [C] person who works to achieve political or social change

be activity /'æktɪvətɪ/ *n* [pl -lɪz]
1 [u] situation in which sth is happening or a lot of things are being done: a lot of - in the street
2 [C, usu pl] something you do for interest or pleasure: leisure/social activities

be actor /'æktə(r)/ *n* [C] (form actress /'æktɪs/) person who acts in plays, films, etc

be actual /'æktʃuəl/ *adj* existing in fact; real **▶** actually *adv* **1** really; in fact: what -ly happened **2** used for showing surprise: He -ly expected me to pay!

acumen /'ækjʊmə/ *n* [u] (form) ability to understand and judge things clearly

acupuncture /'ækjʊpʌŋktʃə(r)/ *n* [u] treatment of illness by sticking small needles into the body

acute /'ækjʊt/ *adj* **1** very great; severe: suffer - hardship **2** very sensitive: an - sense of hearing
▶ a *acute* 'accent *n* [C] mark written over a letter, as over the *e*'s in résumé **▶** a *acute* 'angle *n* [C] angle of less than 90° **▶** acutely *adv*
▶ acuteness *n* [u]

AD (in 'dɪ): *abbr* (in the Christian calendar) since the birth of Jesus Christ

be ad /æd/ *n* [C] (form) = advertisement

adamant /'ædəmənt/ *adj* (form) refusing to change your opinion

Adam's apple /'ædəms 'æpl/ *n* [C] lump at the front of the throat

be adapt /'ædæpt/ *v* **1** [T] make sth suitable for a new use **2** [LT] change your behaviour to deal with a new situation **▶** adaptable *adj* able to change **▶** adaptation /-tʃeɪʃn/ *n* [C, u] result or process of adapting: an -ation of the play for television
▶ adapter *n* [C] kind of electrical plug that allows several plugs to be connected to one socket

be add /æd/ *v* [T] **1** put sth together with sth else: - the flour to the milk

2 put numbers together to get a total **3** say sth more (form) **add up**
1 (inform) seem reasonable; make sense **2** calculate the total of two or more numbers **add up to sth** result in sth; show sth

adder /'ædə(r)/ *n* [C] small poisonous snake

addict /'ædɪkt/ *n* [C] **1** person who cannot stop taking drugs, alcohol, etc **2** person strongly interested in sth or TV - **▶** addicted /'ædɪktɪd/ *adj* - (to) unable to stop taking or using sth **▶** addiction /'ædɪkʃən/ *n* [C, u] **▶** addictive /'ædɪktɪv/ *adj*

be addition /'ædɪʃən/ *n* **1** [u] process of adding numbers together to find their total **2** [C] thing that is added to sth else **(BGM) in addition (to sb/ sth)** as an extra person, thing, etc
▶ additional /-ʃənəl/ *adj* extra
▶ additionally *adv*

additive /'ædɪtɪv/ *n* [C] substance added to sth, esp food

be address /'ædres/ *n* [C] **1** details of where sb lives or works and where letters, etc can be sent: What's your name and -? - on email - **2** formal speech **▶** address *v* [T] **1** write a name and address on sth **2** give a speech to sb

adept /'ædɛpt/ *adj* - (sb/in) skilled at (doing) sth

be adequate /'ædɪkwət/ *adj* enough; satisfactory **▶** adequately *adv*

adhere /əd'hiərə/ *v* [i] (form) - (to) stick firmly to sth (form) adhere to sth (form) act according to a law, rule, etc; follow a set of beliefs
▶ adherence /-rəns/ *n* [u]
▶ adherent /əd'hiərənt/ *n* [C] supporter of a person, group, etc

adhesive /əd'hi:zɪv/ *adj* able to stick to sth **▶** adhesion /əd'hi:zjən/ *n* [u] **▶** adhesive *n* [C, u] substance, eg glue, that makes things stick together

ad hoc /,æd 'hɔ:ɪ/ *adj, adv* made or arranged for a particular purpose; not planned

adjacent /'ædʒəsnt/ *adj* - (to) next to sth

adjective /'ædʒɪktɪv/ *n* [C] (gram) word that describes a noun, eg green in green grass **▶** adjectival /'ædʒɪk'tɪvəl/ *adj*

adjoin /'ædʒɔɪn/ *v* [i] (form) be next to: -ing rooms

adjourn /'ædʒaɪn/ *v* [T] stop a meeting for a time, esp in a court of law **▶** adjournment *n* [C, u]

adjudicate /'ædʒʊdɪkeɪt/ *v* [LT] (form) make an official decision or judgement on sth **▶** adjudication /-tʃeɪʃən/ *n* [u] **▶** adjudicator *n* [C]

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{cl aq} + 2\text{ag aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds Cl^-aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver I^+ ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 $13 \text{Ag}^+\text{aq} + 2\text{F}^-\text{aq} + 2\text{NH}_4^+\text{aq} + \text{Cr}_2\text{O}_7^{2-}\text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7\text{s} + 2\text{NH}_4\text{F}^-\text{aq}$ because two NH_4^+aq

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