

Diane Clement

AT THE TOMATO



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Joy

Diane Clement At The Tomato

Annelies Wilder-Smith



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mole H_2S x 18g H_2O i mole H_2O 149 4g H_2O is your theoretical yield so percent yield $\frac{137}{149} \times 100 = 91.77\%$ 1g H_2O actual yield 149 4g H_2O x 100
91.77 problems 6 8 are similar to problem 5

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