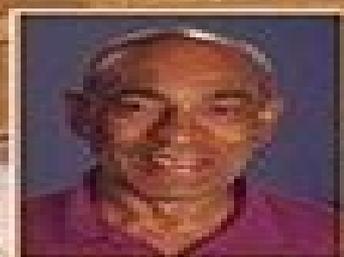
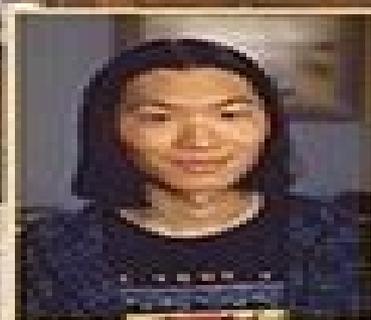
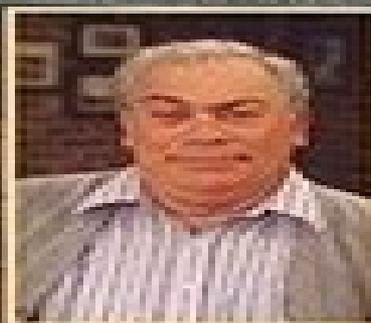


# Crossroads Café

## WORKTEXT A

Crossroads Café



**K. Lynn Savage**  
**Patricia Mooney Gonzalez**  
**Mary McMullin**  
with Kathleen Santopietro Weddel

For  
Middle-Level  
Learners of  
English

# Crossroads Cafe Worktext A English Learning Program

**K. Lynn Savage, Anna Cuomo, Patricia  
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## **Crossroads Cafe Worktext A English Learning Program** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Crossroads Cafe Worktext A English Learning Program**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \frac{7}{2} \text{O}_2(\text{g}) + 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta H = 283 \text{ kJ}$   
g 1  $\text{C}_2\text{H}_6(\text{g})$

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sum of the  $\Delta h$  values of the constituent reactions enthalpy change  $\Delta h$  is independent of the path that a reaction follows to move from reactants

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web thus for reaction 9 2 in which 1 mole of  $H_2O$  and 1 mole of  $OH^-$  combine to form 2 moles of  $H_2O$   $\Delta H = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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