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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h
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reactions and products pb no_3 2aq 2kl aq pb^{2+} 2s $2\text{kno}_3 \text{aq}$ complete ionic equation an equation written with all soluble strong electrolytes shown as ions

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web the ag aq and cl aq ions become agcl s but the na aq ions and the $\text{no}_3 \text{aq}$ ions stay as na aq ions and $\text{no}_3 \text{aq}$ ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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