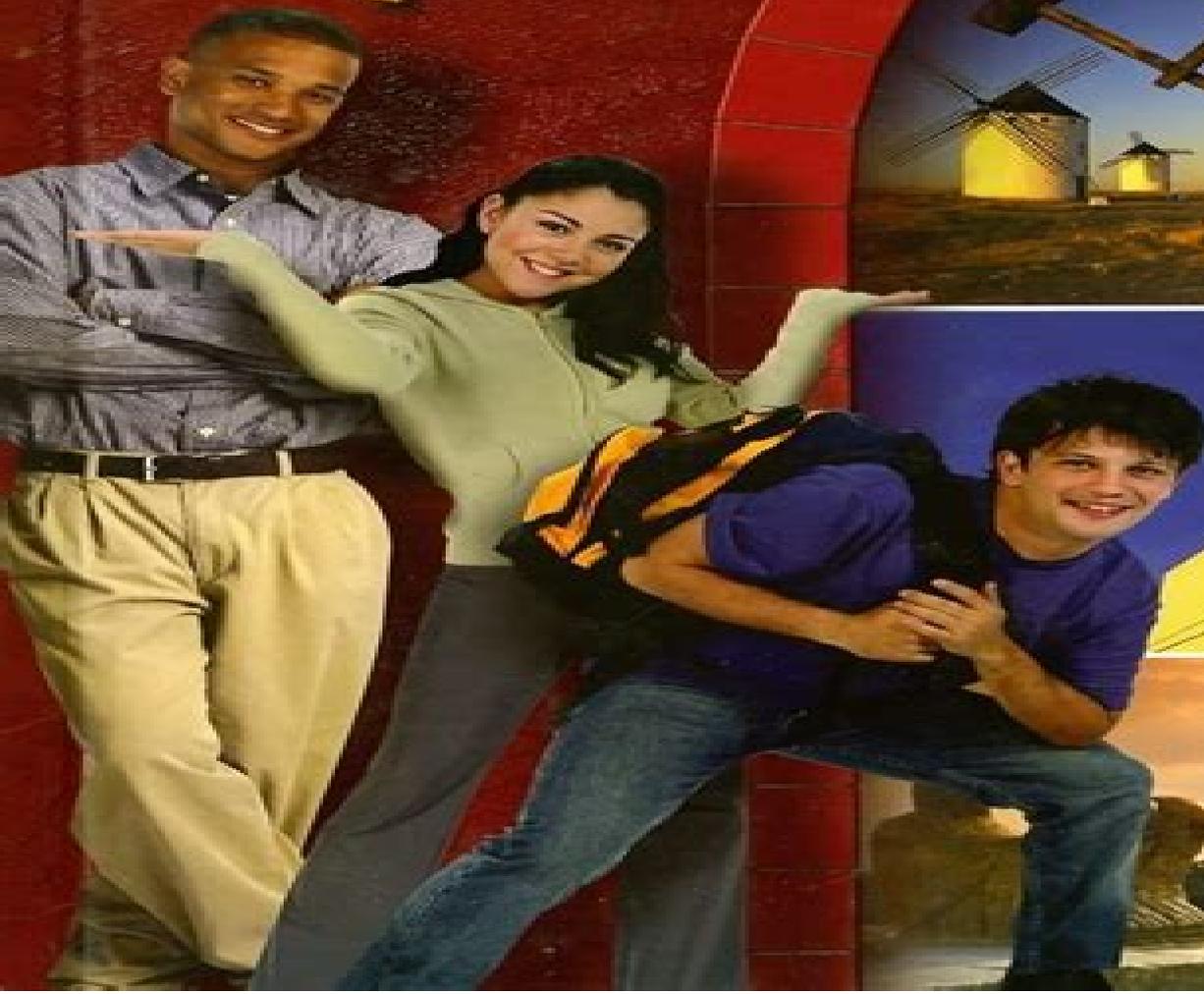


Glencoe Spanish 1

# *¡Buen viaje!*



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 $n_2 + 3h_2 \rightarrow 2nh_3$   
 $mg + oh_2 \rightarrow 2hno_3$   
 $3mg + no_3 \rightarrow 2h_2o + 2nacl_3$   
 $3o_2 + 4al \rightarrow 3o_2 + 2al_2o_3$   
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 $mg + oh_2 \rightarrow mg + oh_2$   
 $hcl \rightarrow hcl$   
 $mgcl_2 \rightarrow mgcl_2$   
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 $o_2 \rightarrow h_2o$   
 note all reactants and products require a coefficient of at least one  
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 metal acid salt hydrogen

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 $2c_2h_2 + 5o_2 \rightarrow 4co_2 + 2h_2o$   
 1 2 0 mol if 1 0 mol of  $c_2h_2$  is burned how many moles of  $co_2$  are formed  
 2 5 mol c if 1 0 mol of  $c_2h_2$  is burned how many moles of oxygen gas are consumed  
 9 a

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