



# Cycling In France

**Xiang Xie**



## **Cycling In France:**

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web in a calorimeter the white solid  $\text{BaSO}_4$  forms and the temperature of the mixture increases to  $28.1 \text{ }^\circ\text{C}$  assuming that the calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is  $4.18 \text{ J/g} \cdot \text{C}$  and that the density of the final solution is  $1.0 \text{ g/ml}$  calculate the enthalpy change per mole

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web problem 8 5 1 6 8 5 1 6 when  $50.0 \text{ g}$  of  $0.200 \text{ M NaCl(aq)}$  at  $24.1 \text{ }^\circ\text{C}$  is added to  $100.0 \text{ g}$  of  $0.100 \text{ M AgNO}_3 \text{ (aq)}$  at  $24.1 \text{ }^\circ\text{C}$  in a calorimeter the temperature increases to  $25.2 \text{ }^\circ\text{C}$  as  $\text{AgCl(s)}$  forms assuming the specific heat of the solution and products is  $4.20 \text{ J/g} \cdot \text{C}$  calculate the approximate amount of heat in joules produced answer