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1 consider a cycle followed by an engine figure 1 to 2 is isothermal 2 to 3 is adiabatic 3 to 1 is adiabatic such a process does  
not exist because choose the correct alternatives 1 a b 2 a c 3 b c 4 c d

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of thermodynamics 1 heat is transferred to a heat engine from a furnace at a rate of 80 mw

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web suppose that the temperature of the water in the previous problem is raised by first bringing it to thermal equilibrium  
with a reservoir at a temperature of 40 c and then with a reservoir at 80 c calculate the entropy changes of a each reservoir b  
of

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thermodynamics 1 a process is carried out between 2 systems it is possible for entropy of one system to decrease true or  
false a true b false view answer

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systems evolve spontaneously towards thermal equilibrium the system s state of maximum entropy in simple terms universe  
entropy the ultimate isolated system only increases and never decreases

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transferred to a heat engine from a furnace at a rate of 80 mw if the rate of waste heat rejection to a nearby river is 50 mw  
determine the net power output for this heat engine a 30 mw

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web oct 9 2023 second law of thermodynamics the entropy of an isolated system not in equilibrium will tend to increase  
over time approaching a maximum value at equilibrium  $\Delta s \Delta q / t \Delta s$  total  $\Delta s$  system  $\Delta s$  surrounding the second law of

thermodynamics introduces the concept of entropy