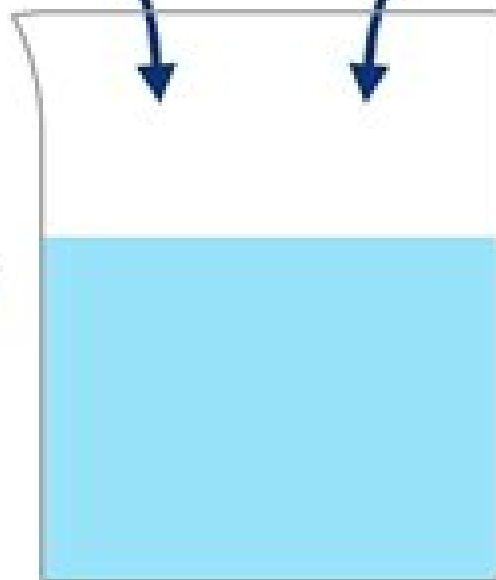




ethanoic acid forms the equilibrium

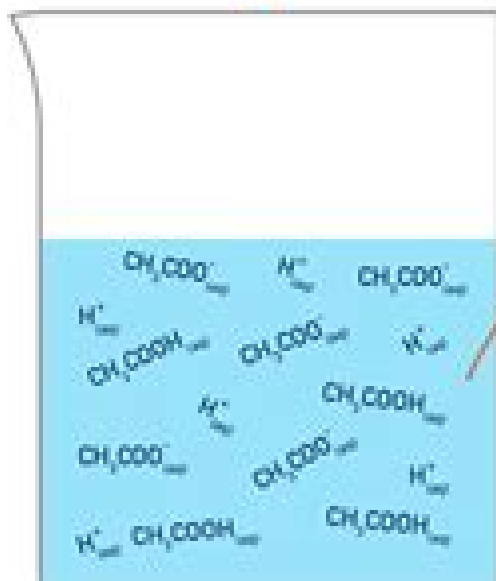


Sodium ethanoate dissolves in water, releasing $\text{CH}_3\text{COO}^-_{(aq)}$ ions into the solution



Mixture of $\text{CH}_3\text{COOH}_{(aq)}$, $\text{CH}_3\text{COO}^-_{(aq)}$ and $\text{H}^+_{(aq)}$ in solution

Buffer solution



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F Rizvi



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