

Chemistry 111 Study Guide Chapters 13-15

Chapter 13

In chapter 13 of the Chemistry 111 textbook, students will learn about the types of chemical reactions and how to write and balance chemical equations. The key concepts that should be studied include:

- The types of chemical reactions (including synthesis, decomposition, single replacement, double replacement, and combustion reactions)*
- How to write and balance chemical equations*
- The factors that affect the rate of a chemical reaction (including concentration, temperature, surface area, and catalysts)*

Important events that should be studied include:

- The discovery of the law of conservation of mass by Antoine Lavoisier*
- The discovery of the law of definite proportions by Joseph Proust*
- The discovery of the law of multiple proportions by Dalton*

Notable figures that should be studied include:

- Antoine Lavoisier*
- Joseph Proust*
- Dalton*

Chapter 14

Key Concepts:

- 1. The role of water in chemical reactions: Water is a solvent and can act as a reactant or product in chemical reactions.*
- 2. The types of intermolecular forces: dipole-dipole, London dispersion, and hydrogen bonding.*
- 3. The relationship between intermolecular forces and the physical properties of substances:*

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