

Roll No: 202400023

Grade: 10

Subject: CHEMISTRY

Total marks: 80

Duration: 3.00hrs

## General Instructions:

Read the following instructions carefully.

There are 39 questions in this all the questions are compulsory

SECTION A question number 1 to 20 multiple-choice questions carrying 1 mark each.

SECTION B question number 21 to 26 very short answer questions carrying 2 marks each.

(30 to 50 words)

SECTION C question number 27 to 33 short answer questions carrying 3 marks each. (50 to 80 words)

SECTION D question number 34 to 36 long answer questions carrying 5 marks each. (80 to 120 words)

SECTION E question number 37 to 39 case-based questions carrying 4 marks each.

## SECTION-A

1. Identify the unsaturated compounds from the following :

(a) Propane (b) Propene (c) Propyne (d) Chloropropane

2. Butanone is a four-carbon compound with the functional group

(a) carboxylic acid (b) aldehyde (c) ketone (d) alcohol

3. A molecule of ammonia ( $\text{NH}_3$ ) has

(a) only single bonds (b) only double bonds

(c) only triple bonds (d) two double bonds and one single bond

4. Vinegar is a solution of

(a) 50%-60% acetic acid in alcohol (b) 5%-8% acetic acid in alcohol

(c) 5%-8% acetic acid in water (d) 50%-60% acetic acid in water

5. Mineral acids are stronger acids than carboxylic acids because (i) mineral acids are completely ionized. (ii) carboxylic acids are completely ionized (iii) mineral acids are partially ionized (iv) carboxylic acids are partially ionized

(a) (i) and (iv) (b) (ii) and (iii) (c) (i) and (ii) (d) (iii) and (iv)

6. Carbon forms four covalent bonds by sharing its four valence electrons with four univalent atoms, e.g., hydrogen. After the formation of four bonds, carbon attains the electronic configuration of

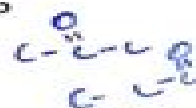
(a) helium (b) neon (c) argon (d) krypton

7. Chlorine reacts with saturated hydro-carbons at room temperature in the

(a) absence of sunlight (b) presence of sunlight

(c) presence of water (d) presence of hydrochloric acid

8. In the soap micelles



# Chemistry Split

**SB Merriam**



## **Chemistry Split:**

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