

Clinical Chemistry

- **NORMAL VALUES** → ^(basic) determined increased or decreased level of chemicals.
- **HYPD** → Decrease in values. • **HYPER** → increase in values.
- **STANDARD** → the basis for determining the concentration of the analyte.
- **CONTROL** → The basis for determining the accuracy of the test.

a. Normal b. Abnormal! } → a guide to determine whether the

test is reliable or not.

S/S: polyphagia, polydipsia, polyuria, Prolonged healing of wounds

Tentative Dx → Diabetes Mellitus (Diabetes)

↳ Inability of the body to produce insulin.

Concentration of sugar/carbohydrate
N.V: 0.6 - 0.9 mmol/L

INDICATION OF REACTIONS IN CHEMISTRY
@ 37°C

ANT increased value

Standard 10 ml 10 ml 10 ml

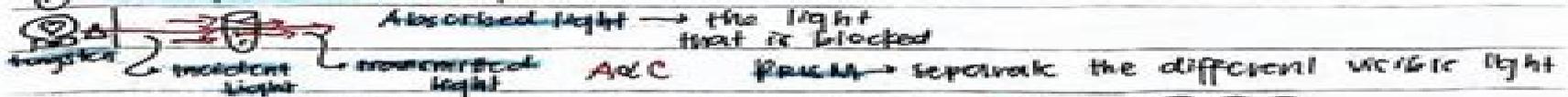
Standard 0.325 0.658 0.311 0.962

Reaction 0.325 0.658 0.325 0.325

$A = \frac{0.862}{0.325} \times 6 = 15.91 \text{ mmol/L (increased value) } \checkmark$
 $T = \frac{0.658}{0.325} \times 6 = 12.14 \text{ mmol/L (value of test) HYPERGLYCEMIA}$
 $N = \frac{0.311}{0.325} \times 6 = 5.74 \text{ mmol/L (normal value) } \checkmark$

METHODS USED IN CL ANALYSIS

1. SPECTROPHOTOMETRY



- * Darker the analyte, darker the concentration. → ☹️ ☹️ ☹️ ☹️ ☹️
- * If it is darker, it absorbed more light.
- * **Absorbance** → directly proportional to the concentration of the analyte
- * **Transmittance** → inversely proportional to the concentration.

2. NEPHELOMETRY

↳ we measure the scattered light.

3. TURBIDIMETRY

↳ we measure the concentration of analyte based on its turbidity.

Clinical Chemistry An Overview

Daniela Niemeyer



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