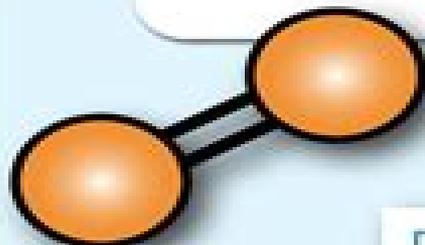


Chemistry of Life

BUNDLE



Organic compounds
Living things are made up of organic compounds (also known as macromolecules). These always have carbon-hydrogen bonds. Carbon makes up the backbone, due to it being a relatively stable atom because it can bond to other carbon atoms.

Functional groups
Functional groups are chemical patterns/groups that are added to hydrocarbons to give the macromolecule particular properties that allow it to have a certain function. They are often R-groups. Here are 7 types:

Amino	Phosphate
$-NH_2$	$-PO_4$
$R-N$	$R-O-P$

Bonds & molecules
A bond is a linking attraction between two or more atoms, ions or molecules. It can only be broken by chemical reactions. When two or more atoms bond, they form a molecule. This can be atoms of the same element, or different atoms to form a compound.

Diatomic molecules
A molecule of oxygen, O_2

Triatomic molecules
A molecule of water, H_2O

Diatomic molecules
A molecule of nitrogen, N_2

Triatomic molecules
A molecule of oxygen, O_3

Chemistry of Life
BASIC CHEMISTRY REVIEW

atoms and life
All living things are made of atoms. An atom is the smallest part of an element. Many different types of atoms exist and they are arranged in the Periodic table of elements. Atoms are made up of subatomic particles.

Label the structure of this oxygen atom

Electron shell, Electron, Neutron, Proton, Nucleus

	Proton	Neutron	Electron
Charge	+1	0	-1
Mass	1	1	1/1836

How many electrons are in the first shell? 2
How many electrons are in the outer shell? 6
What do we call electrons in the outer shell? valence electrons

mixtures
A mixture is a mix of 2 or more different types of atoms that are not chemically bonded. They can be separated by physical means.

compounds
A compound is made of 2 or more different types of atoms that are chemically bonded. They are not separated by physical means.

elements
An element is a substance made of one type of atom. It can be broken down into atoms.

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