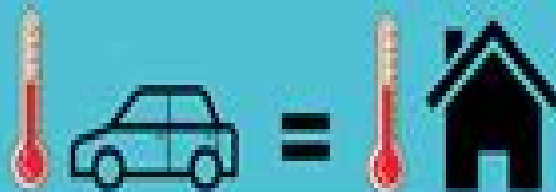


# Laws of Thermodynamics

## Zeroth Law

Defines temperature equality: if two bodies are each with a third, they're in equilibrium.



## First Law

Energy remains constant; it transforms but cannot be created or destroyed.



Total Energy  
Equal



## Second Law

Entropy increases in isolated systems; heat flows from hot to cold.



## Third Law

Entropy approaches a constant value as temperature approaches absolute zero.



# Basic Thermodynamics

**N Noddings**



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