



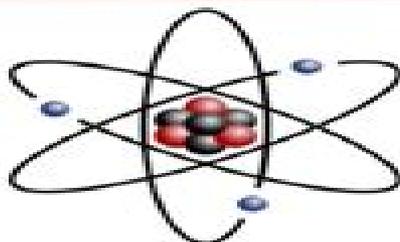
Monares, Julius Ceazar A.

Science Technology Engineering Mathematics

## Properties of Matter

- Characteristics that describe a sample of matter are called properties.
- The mass of an object is a measure of the amount of matter the object has.
- Each sample of matter occupies space.
- The measure of the space occupied by an object is called volume.
- All matter have mass and volume. There are other properties of matter such as hardness, texture, color, flexibility, malleability, and electrical conductivity which vary from one sample to another.

## Discovery of Atoms



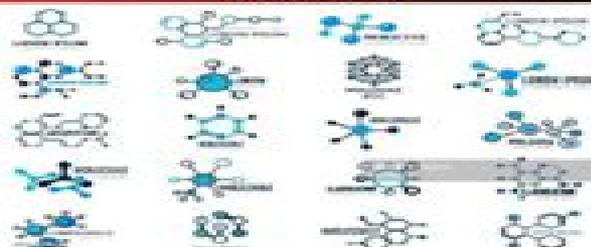
- The ancient Greek philosophers proposed ideas about what matter was made of.
- Almost 2,500 years ago, Leucippus and his disciple, Democritus believed that nature consisted of two things, "atoms and the void that surrounds them" (Knieram, 1995-2013).
- They believed that "atoms are physically, but not geometrically, indivisible."
- For Democritus, atoms are indestructible and completely full, so there is no empty space.
- Both Leucippus and Democritus had the idea that there are many different kinds of atoms and each of them had specific shape and size and that all atoms move randomly around in space.
- Democritus believed that any piece of matter can be divided and subdivided into very small particles but that this process ended at some point when a piece is reached that could not be further divided.
- He called this particle, atomos, a Greek word which means indivisible particle. Democritus' ideas about the atom were later challenged by other Greek philosophers, most strongly by Aristotle.

## John Dalton



- John Dalton presented concrete evidence that all matter is made of very small particles called atoms.
- An atom is the smallest particle of an element that has all the properties of the element.
- Atoms of most elements have the ability to combine with other atoms.
- Different elements have different properties because the combining atoms are different and the way the atoms are joined together are different.

## Molecules



- A molecule is a particle consisting of two or more atoms combined together in a specific arrangement.
- It is an electrically neutral particle.
- It is the smallest particle of an element or compound that can exist independently.
- For example, a molecule of water consists of an oxygen atom combined with two hydrogen atoms.
- Atoms of the same element can also combine to form a molecule.
- For example, oxygen in the air consist of oxygen molecules which are made up of two oxygen atoms.

## Size of an Atom

- Atoms are too small to observe. These particles cannot be seen under the high-powered light microscopes used in school laboratories.
- The size of an atom is measured in angstroms.
- One angstrom is a unit of length equal to one ten millionth of a millimetre.

# Chemistry For Biosciences

**Léo Pomar**



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