

CHEMISTRY LESSON PLAN:

The Structure of the Atom, Chemical Bonding

NAME: RICKY E. BAHGY

DATE:

YEAR/COURSE/MAJOR/BLOCK: BEED -2 BLOCK 1

INSTRUCTOR:

Content

Understand and list the properties of protons, neutrons, and electrons.

Key Words: atom, nucleus, neutron, electron, proton, atomic number, mass number

-Ionic/ covalent bonds

Goals / Aims of The Lesson Plan

1. Students will understand the core parts of an atom.
2. Students will know the difference between an element and compound.
3. Students will be able to differentiate between elements of atoms based on their atomic composition.
4. To draw 'dot and cross' diagrams
5. To identify ionic or covalent compounds.

Objectives

1. Summarize Rutherford discovery
2. Explain the differences between isotopes and physical activity that it demonstrates.
3. Solve problems involving mass in grams
4. To list properties of ionic compounds
5. To list properties of covalent compounds.

Materials and Aids

Graphing calculator, textbook, Periodic Table

Procedure

A. Introduction-

1. We will discuss Cathode Rays and electrons; Gold foil experiment
2. Composition of Nucleus
3. Isotopes - How are isotopes formed and how stable are they?
4. Demo Time: A synthesis reaction between Hydrogen and Oxygen
5. A combustion reaction showing how water breaks apart to make $2H + O$

B. Development-

Chemistry Lesson Plans

Peipei Pang



Chemistry Lesson Plans:

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